CONCEPT MAP

SPECIFICATION OF CONFIGURATION

 The configuration with —OH group at right side is given D-configuration arbitrarily whereas the configuration with —OH group at left side is given L-configuration.

Chiral molecules which can be chemically related to *D*-glyceraldehyde were designated as *D*-isomer and the molecules related

CHO CHO
$$H \longrightarrow OH$$
 HO $\longrightarrow H$
 CH_2OH CH_2OH
 $D(+)$ -Glyceraldehdye $L(-)$ -Glyceraldehdye

to L-glyceral dehyde are designated as L-isomer.

 In polyhydroxy compounds (sugars) having more than one chiral center, the configuration of stereocenter farthest from carbonyl group is compared with glyceraldehyde configuration.

Rules for Assigning Configuration

STEP I: Order of Priority

Rule 1: If all the four atoms attached to chirality centre are different, their priority numbers are decided on the basis of their higher atomic numbers.

$$e.g., -I > -Br > -Cl > -F > -O > -N > -C > -H$$

Rule 2: If two of the atoms attached to chirality centre are same then the relative priorities can be decided by comparing the second (or even third) atom of the respective group.

$$^{3}\text{CH}_{3}$$
 $^{2}\text{CH}_{2}\text{CH}_{3}$ $^{2}\text{CH}_{2}\text{CH}_{3}$ $^{2}\text{CH}_{2}\text{CH}_{3}$ $^{3}\text{CH}_{3}$ ^{4}H ^{1}Br ^{1}Br $^{2}\text{Butan-2-ol}$ $^{2}\text{Bromobutane}$

Rule 3: This is an extension of rule II. If the second atoms is attached by double bond or triple bond, then double bonds are treated as they have two single bonds and triple bonds as they have three single bonds *i.e.*,

$$>$$
C=O equals to $>$ C-O and $-$ C \equiv N equals to $-$ C $\stackrel{N}{\leq}$ N

Thus, priority order is $C_6H_5 \rightarrow -C \equiv CH \rightarrow -CH = CH_2 \rightarrow -CH(CH_3)_2$

STEP II

After deciding the priority of four atoms or groups attached to the chirality centre, the molecule (formula) is now rotated so that the group of lowest priority (4) is directed away from the eye (viewer). The arrangement of remaining three groups is viewed in decreasing order of their priorities, *i.e.*, from 1 to 2 to 3. In looking so, if the eye travels in a clockwise direction, the configuration is specified as R (from rectus Latin for right), while in case if eye travels in the anticlockwise direction, the configuration is specified as S (from sinister Latin for left).

Relative Configuration

The configuration of the (OH) group at penultimate carbon of various stereoisomers are studied with respect to glyceraldehyde.

CONFIGURATION

The arrangement of atoms or groups in space that characterises a stereoisomer is called its configuration. The sign of rotation of a stereosiomer is not related to its configuration.

Absolute Configuration

Cahn-Ingold-Prelog system has been developed to assign absolute configuration. If the lowest priority group occupies horizontal position (left or right) in the Fischer projection formula, then change the configuration obtained from (R) to (S) or (S) to (R). For example,

Golden Rule

Assigning R, S Configuration in Fischer Projection

 When atom or group of lowest priority is at the bottom of vertical line:

There is an easy way of assigning *R* and *S* configuration to optical isomers represented by Fischer projection formula. Firstly assign priorities to the groups/atoms attached to chiral centre without making any change/interchange in the strength. Find out the configuration. Now if the lowest priority group occupies vertical (upward or downward) position in the original Fischer projection, then the configuration obtained above gives the actual configuration of the molecule. For example,

Assigning R, S configuration of the compounds with more than one chiral centres

If a compound has two or more chiral carbon atoms, separate configuration is assigned to each carbon atom on the basis of sequence rule. In such compounds, application of golden rule is more convenient.

For example,

$$^{1}COOH$$
 $^{(4)}H$
 $^{(4)}H$
 $^{(3)}OH$
 $^{(1)}HO$
 $^{(4)}H$
 $^{(4)}H$
 $^{(3)}OH$
 $^{(4)}H$
 $^{(4)}H$

When atom or group of lowest priority is not on the vertical line:

If the lowest priority group/atom does not lie at vertical downward or upward position (bottom or above), then make two interchanges of atoms or groups in such a way that the atom or group of lowest priority is placed either vertically upward or downward. Now, assign the configuration according to path followed in decreasing order of priority.

Effect on the configuration

By exchanging a horizontal and vertical ligand :

$$CH_3$$
 $Br \xrightarrow{S \to R} H_3 \xrightarrow{S} \xrightarrow{Br} \begin{bmatrix} H \text{ on vertical line;} \\ Clockwise and thus } (R) \end{bmatrix}$
 C_2H_5
 C_2H_5
 C_2H_5

By 180° vertical or horizontal rotation outside the plane of paper: This
type of operation is not allowed in Fischer projection because any flipping
out of the plane of paper (side to side or top to bottom) would change
the ligands formerly projecting behind the plane of paper to projecting
towards the observer.

Those ligands formerly pointing towards the observer would change to behind the paper.

By a 90° rotation on the plane of paper :

This type of operation is also not allowed in Fischer projection. For example,

$$H \xrightarrow{CH_3} Br \longrightarrow H_3C \xrightarrow{Br} C_2H_5$$
 (I)
 $H \xrightarrow{(I')} C_2H_5$

In (I), H and Br project behind the plane, but a 90° rotation on the plane does not change the projection of bonds towards or away from the observer.